BS ISO 14388-2:2014



## **BSI Standards Publication**

# Soil quality - Acid-base accounting procedure for acid sulfate soils

Part 2: Chromium reducible sulfur (CRS) methodology



This British Standard is the UK implementation of ISO 14388-2:2014.

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A list of organizations represented on this committee can be obtained on request to its secretary.

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# Soil quality - Acid-base accounting procedure for acid sulfate soils —

Part 2:

## **Chromium reducible sulfur (CRS)** methodology

Qualité de l'eau — Méthode de comptage acide-base pour les sols sulfatés acides —

Partie 2: Méthode de sulfato réduction au chrome



BS ISO 14388-2:2014 **ISO 14388-2:2014(E)** 

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#### **Foreword**

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The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

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For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT) see the following URL: Foreword - Supplementary information

The committee responsible for this document is ISO/TC 190, *Soil quality*, Subcommittee SC 3, *Chemical methods and soil characteristics*.

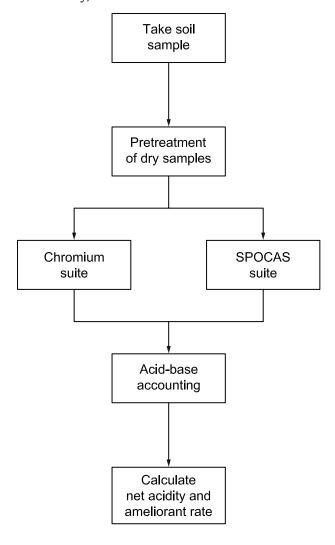
ISO 14388 consists of the following parts, under the general title *Soil quality — Acid-base accounting procedure for acid sulfate soils*:

- Part 1: Introduction and definitions, symbols and acronyms, sampling and sample preparation
- Part 2: Chromium reducible sulfur (CRS) methodology
- Part 3: Suspension peroxide oxidation combined acidity and sulfur (SPOCAS) suite analysis

#### Introduction

The objective of this part of ISO 14388 is to determine the net acidity (or alkalinity) of acid sulfate soils by providing a streamlined approach for determination of the various components of soil acidity and/or alkalinity, depending on pH. The chromium suite combines the measurement of chromium reducible sulfur ( $S_{CR}$ ) with various measures of existing acidity and acid neutralizing capacity (ANC) using a decision-tree based on the value of pH<sub>KCl</sub> (Figure 1) as the basis for determining an acid-base account for acid sulfate soils.

The results required to determine net acidity vary with the soil's actual acidity, as represented by  $pH_{KCl}$ . Table A.1 of ISO 14388-1 shows the analyses required for the Chromium suite. This table uses results reported in acidity units. Alternatively, results in sulfur units can be utilized.



#### Key

- a acidity titration
- b sulfur determination
- c acid neutralising determination
- d calculated parameter

Figure 1 — Chromium suite flow diagram

## Soil quality - Acid-base accounting procedure for acid sulfate soils —

#### Part 2:

### Chromium reducible sulfur (CRS) methodology

WARNING — Persons using this part of ISO 14388 should be familiar with usual laboratory practice. This part of ISO 14388 does not purport to address all of the safety problems, if any, associated with its use. It is the responsibility of the user to establish appropriate safety and health practices and to ensure compliance with any national regulatory conditions.

#### 1 Scope

This part of ISO 14388 specifies a suite of methods used to determine the net acidity in acid sulfate soils. This part of ISO 14388 specifies a method for measuring chromium reducible sulfur ( $S_{CR}$ ) by iodimetric titration of distilled hydrogen sulfide trapped as zinc sulfide, following acidic chromous chloride digestion. This method determines inorganic sulfides (e.g. pyrite, marcasite, greigite, mackinawite) and elemental sulfur in acid sulfate soil without interferences from organic sulfur and oxidized forms of sulfur such as sulfate.

On a separate test portion of soil, the pH in a 1 mol/l KCl soil suspension (pH<sub>KCl</sub>) is determined. When pH<sub>KCl</sub> is < 6,5, titratable actual acidity (TAA) is then determined. Subsequently, potassium chloride extractable sulfur (S<sub>KCl</sub>), calcium (Ca<sub>KCl</sub>), and magnesium (Mg<sub>KCl</sub>) can also be determined. Where jarosite is identified in the soil (or where pH<sub>KCl</sub> is < 4,5), net acid-soluble sulfur (S<sub>NAS</sub>) is determined by the difference between hydrochloric acid extractable sulfur (S<sub>HCl</sub>) and potassium chloride extractable sulfur. On samples where pH<sub>KCl</sub> is < 6,5, acid neutralizing capacity is determined by measuring either inorganic carbon (C<sub>IN</sub>) by combustion furnace, or ANC<sub>BT</sub> (ANC measured by back-titration of acid remaining following an acid digest).

#### 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 385-1, Laboratory glassware — Burettes — Part 1: General requirements

ISO 648, Laboratory glassware — Single-volume pipettes

ISO 835-1, Laboratory glassware — Graduated pipettes — Part 1: General requirements

ISO 835-2, Laboratory glassware — Graduated pipettes — Part 2: Pipettes for which no waiting time is specified

ISO 835-3, Laboratory glassware — Graduated pipettes — Part 3: Pipettes for which a waiting time of 15 s is specified

ISO 1042, Laboratory glassware — One-mark volumetric flasks

ISO 1770, Solid-stem general purpose thermometers

ISO 3696, Water for analytical laboratory use — Specification and test methods