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C40-A

Analytical Procedures for the Determination of Lead in Blood and Urine; Approved Guideline



This document provides guidelines for the measurement of lead in blood and urine, including specimen collection, measurement by graphite furnace atomic absorption spectrometry (GFAAS) and anodic stripping voltammetry (ASV), quality assurance, and quality control.

A guideline for global application developed through the Clinical and Laboratory Standards Institute consensus process.

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Analytical Procedures for the Determination of Lead in Blood and Urine; Approved Guideline

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Abstract

Analytical Procedures for the Determination of Lead in Blood and Urine; Approved Guideline (CLSI document C40-A) is intended for use by the clinical laboratory testing community involved in the collection and measurement of lead in blood and urine. The guideline addresses the clinical significance of lead measurements, specimen collection, lead determination by graphite furnace atomic absorption spectrometry and anodic stripping voltammetry, reference materials, quality control procedures, and laboratory policy.

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Contents

Abstract..... v

Committee Membership..... v

Active Membership..... vii

Foreword..... xv

1 Introduction..... 1

2 Scope..... 1

3 Definitions 2

4 Clinical Significance of Lead Measurements 2

 4.1 Absorption of Lead and Its Internal Distribution Within the Body 2

 4.2 Toxic Effects of Exposure to Lead in Children and Adults..... 3

 4.3 Concentration of Lead in Blood Deemed Safe for Children/Adults..... 5

 4.4 Use of Blood Lead Measurements as a Marker of Lead Exposure 6

 4.5 Use of Urinary Lead Measurements in the Mobilization Test..... 6

 4.6 *In vivo* X-ray Fluorescence Measurements of Bone Lead..... 7

 4.7 Use of Hair, Sweat, and Other Tissues/Fluids..... 7

 4.8 Erythrocyte Protoporphyrin Analysis and Its Use 7

 4.9 Reference Intervals for Lead in Blood and Urine..... 8

5 Analytical Methods for Lead 9

 5.1 Early Methods for the Determination of Lead..... 9

 5.2 Flame AAS and MIBK Extraction 9

 5.3 Delves-Cup FAAS Method 9

 5.4 Anodic Stripping Voltammetry (ASV)..... 10

 5.5 Graphite Furnace Atomic Absorption Spectrometry (GFAAS) 10

 5.6 Mass Spectrometric Methods 11

 5.7 Future Approaches to Blood Lead Measurements..... 12

6 Procedure for Collecting Blood and Urine for Lead Determination (see Section 10.3) 13

 6.1 Procedure for Collecting Capillary Blood Specimens..... 13

 6.2 Procedure for Collecting Venous Blood Specimens 17

 6.3 Procedure for Collecting Urine Specimens 19

7 Procedure for Pb in Blood and Urine by Graphite Furnace Atomic Absorption Spectrometry (GFAAS)..... 21

 7.1 Principle of Measurement and Method Summary 21

 7.2 Instrumentation Requirements..... 21

 7.3 Reagents and Other Materials..... 24

 7.4 Preparation of Reagents, Standards, and Specimens for Analysis 25

 7.5 Furnace Operation and Maintenance 28

 7.6 Calculations 29

 7.7 Troubleshooting..... 31

8 Procedures for Pb in Blood by Anodic Stripping Voltammetry (ASV)..... 32

 8.1 Method Principle 32

Contents (Continued)

8.2	Apparatus.....	33
9	Reference Materials and Quality Control for Lead in Blood and Urine	35
9.1	Reference Materials.....	35
9.2	Quality Control.....	40
10	Laboratory Policy	42
10.1	Laboratory Certification and Proficiency Testing	42
10.2	Follow-up and Rescreening.....	42
10.3	General Policies for Lead Analyses (See Section 6)	43
	References.....	46
	Appendix A. Procedure for Checking Materials and Specimen Collection Supplies for Lead Contamination.....	56
	Appendix B. Laboratory Certification and Proficiency Testing in the United States.....	59
	Appendix C. Reporting Practices, Risk Classifications, and Rescreening Timetables Recommended for Blood Lead Results in the United States	77
	Summary of Comments and Subcommittee Responses.....	81
	Related NCCLS Publications.....	92

Foreword

The primary impetus for this document was the pivotal statement on childhood lead poisoning prevention published by the U.S. Centers for Disease Control and Prevention (CDC) in 1991 which lowered the concentration of lead in blood (BPb) deemed harmful to children from 25 $\mu\text{g/dL}$ to 10 $\mu\text{g/dL}$ (0.48 $\mu\text{mol/L}$).¹ At this lower BPb level, a widely used biochemical screening test for lead exposure, the erythrocyte protoporphyrin (EP) test, became redundant because EP is insensitive to such low-level lead exposure. The statement recommended that EP screening for lead exposure be discontinued in favor of a direct BPb test. The worldwide impact of the 1991 CDC statement is still being felt as private and public health laboratories continue to respond to the need for inexpensive BPb testing on small capillary blood specimens.

In November 1997, CDC updated its recommendations for screening young children for lead poisoning.² The CDC document addresses specific concerns about the extent to which universal screening of all children should be, or can be, implemented. As part of the release of the 1997 document, CDC is providing specific advice and materials to blood lead laboratories (see Appendix C in the CDC document) that complement the guidelines proposed in C40-A. These materials may be downloaded from the CDC's worldwide website at the following URL address:

<http://www.cdc.gov/nceh/lead/lead.htm>

Because CDC recommendations are often adopted internationally, updated tables from the 1997 document as well as appropriate follow-up actions for confirmed, elevated, blood-lead levels are included here in Appendix C. Information on laboratory accreditation in the U.S. is also provided in Appendix B, along with details of proficiency testing (or external quality assessment) programs for blood lead in the U.S., Canada, and the European Union.

In 1991, established proficiency testing requirements for BPb accuracy were tightened to reflect the improvements in current analytical methodology and the lower concentrations of BPb deemed harmful. Some laboratories using older methods for BPb were unable to maintain proficiency, and were required to improve their analytical method performance. Many were understandably concerned that the analytical technology for making accurate, contamination-free measurements of low levels of lead in capillary blood specimens did not exist. In the years since the 1991 CDC statement was released, it has been shown that current analytical methods can easily measure BPb concentrations below 10 $\mu\text{g/dL}$ with good accuracy and precision. Analytical accuracy continues to improve as evidenced by the performance of participating laboratories in numerous quality assurance and proficiency testing programs. Many laboratories have succeeded in setting up the BPb analysis in-house.

This document addresses many concerns and questions that laboratories have about BPb measurements. Issues of accuracy, interferences, contamination control, and troubleshooting affect all analytical methods. Two analytical methods for blood lead are in routine use at the current time: graphite furnace atomic absorption spectrometry (GFAAS) and anodic stripping voltammetry (ASV). Instrumentation for GFAAS is available from many commercial sources, and a recommended analytical method is described in some detail here. Commercial ASV instrumentation specifically for the BPb analysis is currently available from a single manufacturer. A detailed ASV procedure, which includes use of a proprietary reagent, is provided by the manufacturer. In keeping with NCCLS policy, details of the commercial ASV method are *not* duplicated here; rather, the procedure is summarized, and specific recommendations are given that can help with troubleshooting performance problems. Several other methods are reviewed for their application to BPb measurements, including Delves-cup microsampling flame AAS and inductively coupled-plasma mass spectrometry (ICP-MS). As a relatively high-cost, multielement technique, ICP-MS is unlikely to be used for BPb measurements in most routine situations.

Foreword (Continued)

Analytical chemistry is an evolving science, with new innovative technologies appearing continually. It is quite likely that new, inexpensive, portable technologies for BPb will appear in the future. While it is not possible to address these emerging technologies in the current document, they will no doubt be included in future revisions.

In producing C40-A, the intention of the NCCLS Subcommittee on Lead was to reach a consensus, so an approved guideline can be distributed to laboratories that carry out lead determinations in blood and urine specimens. The subcommittee also intended for the document to be broad in perspective and, thus, an educational resource for laboratorians.

Standard Precautions

Because it is often impossible to know what might be infectious, all human blood specimens are to be treated as infectious and handled according to "standard precautions." Standard precautions are new guidelines that combine the major features of "universal precautions and body substance isolation" practices. Standard precautions cover the transmission of any pathogen and thus are more comprehensive than universal precautions which are intended to apply only to transmission of blood-borne pathogens. Standard precaution and universal precaution guidelines are available from the U.S. Centers for Disease Control and Prevention (*Guideline for Isolation Precautions in Hospitals*. Infection Control and Hospital Epidemiology. CDC. 1996;Vol 17;1:53-80.), [MMWR 1987;36(suppl 2S):2S-18S] and (MMWR 1988;37:377-382, 387-388). For specific precautions for preventing the laboratory transmission of blood-borne infection from laboratory instruments and materials; and recommendations for the management of blood-borne exposure, refer to NCCLS document M29—*Protection of Laboratory Workers from Instrument Biohazards and Infectious Disease Transmitted by Blood, Body Fluids, and Tissue*.

Key Words

Analysis, anodic stripping voltammetry, blood, electrothermal atomic absorption spectrometry, graphite furnace, lead poisoning, quality control, reference materials, urine

Abbreviations

AAS	Atomic absorption spectrometry
APDC	Ammonium pyrrolidine dithiocarbamate
ASV	Anodic stripping voltammetry
BPb	Blood lead
CDC	Centers for Disease Control and Prevention
CRM	Certified reference material
CV	Coefficient of variation (synonymous with RSD)
DL	Detection limit
DMSA	2,3-dimercaptosuccinic acid
EDL	Electrodeless discharge lamp
EDTA	Ethylenediaminetetraacetic acid
EP	Erythrocyte protoporphyrin
ETAAS	Electrothermal atomic absorption spectrometry
FAAS	Flame atomic absorption spectrometry
FPP	False-positive proportion
GFAAS	Graphite-furnace atomic absorption spectrometry
HCL	Hollow-cathode lamp
ICP-MS	Inductively coupled plasma mass spectrometry

Abbreviations (Continued)

ID	Isotope dilution
IFCC	International Federation of Clinical Chemistry
ISO	International Organization for Standards
IUPAC	International Union of Pure and Applied Chemistry
LC	Liquid chromatography
LDC	Lowest determinable concentration
MIBK	Methylisobutylketone
m_0	Characteristic mass
NHANES	National Health and Nutrition Examination Surveys
NIST	National Institute of Standards and Technology
NRSCL	National Reference System for the Clinical Laboratory
PT	Proficiency test(ing)
OSHA	Occupational Safety and Health Administration
QC	Quality control
RM	Reference material
RSD	Relative standard deviation (synonymous with CV)
SD	Standard deviation
SI	Système International d'Unités
SRM	Standard reference material
STPF	Stabilized temperature platform furnace
TCA	Trichloroacetic acid
TIMS	Thermal ionization mass spectrometry
UPb	Urine lead
ZPP	Zinc protoporphyrin

Analytical Procedures for the Determination of Lead in Blood and Urine; Approved Guideline

1 Introduction

The primary aim of this document is to provide clinical laboratories with concise guidelines for measuring lead in blood and urine. It includes specimen collection and measurement by the two principal analytical methods currently in routine use: electrothermal atomic absorption spectrometry (ETAAS), also widely known as graphite furnace atomic absorption spectrometry (GFAAS); and anodic stripping voltammetry (ASV). The document also includes guidelines for quality assurance, quality control, and information on proficiency testing programs and laboratory certification. In developing this guideline the subcommittee recognizes that a single-standard method for GFAAS may not be possible given the complexity of the analytical instrumentation required and the different ways in which manufacturers implement similar features. Furthermore, there is currently only one commercial source of ASV equipment available specifically for BPb measurements. The analyst is free to choose which technique best suits the laboratory's needs, and may modify the recommended procedure to achieve successful analyses. However, whether following the recommended procedure or a modified version, the analyst is responsible for ensuring that the procedure adopted in the laboratory is validated as described under the relevant section.

2 Scope

This document is provided for all laboratories attempting the determination of lead in blood or urine. Laboratories new to the analysis will benefit from the many years of experience accumulated in the laboratories of the committee members, advisors, and observers. A background section on the clinical significance of lead measurements is included to help laboratorians and others understand the context in which these measurements are made. Recommended procedures for collecting blood (both capillary and venous) and urine specimens are given. A detailed analytical procedure is recommended for use with GFAAS equipment. Since an ASV method for BPb is provided by the equipment manufacturer, a detailed procedure is not duplicated here but is summarized. Furthermore, some useful information is included in the ASV section to help users avoid performance problems. Other analytical methods are referenced either within a historical context (e.g., Delves-cup AAS) or with respect to specialized applications (e.g., ICP-MS).

The International Federation of Clinical Chemistry (IFCC) and the International Union of Pure and Applied Chemistry (IUPAC) recommend the exclusive use of the liter as unit of volume when reporting laboratory results using the Syst \bar{I} me International d'Unit \bar{e} s (SI). Many published BPb values are universally reported as mass concentration (unit $\mu\text{g/dL}$, $\mu\text{g}/100\text{ mL}$) rather than as substance concentration (unit $\mu\text{mol/L}$). Pediatric urine lead (UPb) measurements are also reported as mass concentration (unit $\mu\text{g/L}$) and, given the total volume of urine collected, the total lead excreted is calculated and reported (unit μg). In this document, the units for BPb are $\mu\text{g/dL}$; the units for UPb are $\mu\text{g/L}$. The IUPAC/IFCC-recommended units of substance (elemental) concentration ($\mu\text{mol/L}$, μM) are included in parentheses, where appropriate. Conversion factors are given below:

$$\text{blood lead } (\mu\text{g/dL}) \times 0.04826 = \text{blood lead } (\mu\text{mol/L})$$

$$\text{urine lead } (\mu\text{g/L}) \times 0.004826 = \text{urine lead } (\mu\text{mol/L})$$